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Step 1 of 1 In the given case, the 5 kg cube of warm iron (60 o C) put thermal contact with 10 kg cold iron (15 o C). The molecules in hot body have more kinetic energy than molecules in cold body. In the molecular collisions, the energy is transfer from molecules of hot body to molecules of cold body.

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Solutions for Problems in Chapter 14 is solved. 1CQ; 1E; 2CQ; 2E; 3CQ; 3E; 4CQ; 4E; 5CQ; 5E; 6CQ; 6E; 7CQ; 7E; 8CQ; 8E; 9CQ; 9E; 10CQ; 10E; 11CQ; 11E; 12CQ; 12E; 13E; 14E; 15E; 16E; 17E; 18E; 19E; 20E; 21E; 22E; 23E; 24E; 25E; 26E; 27E; 28E; 29E; 30P; 31P; 32P; 33P; 34P; 35P; 36P; 37P; 38P; 39P; 40P; 41P; 42P; 43P; 44P; 45P; 46P; 47P; 48P; 49P; 50P; 51P; 52P; 53P; 54P; 55P; 56P; 57P; 58P; 59P; 60P; 61P; 62P; 63P; 64P; 65P; 66P; 67P; 68P; 69P; 70P; 71P; 72P; 73P; 74P; 75CP; 76CP; 77CP; 78CP ...

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1.1 Mathematics and Physics pages 3–10 page 10 13. Math Why are concepts in physics described with formulas? The formulas are concise and can be used to predict new data. 14. Magnetism The force of a magnetic field on a charged, moving particle is given by F Bqv, where F is the force in kg m/s2, q is the charge in A s, and v is the speed in m/s.

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14. The time on any of these clocks is a straight-line function of that on another, with slopes $\mu = 1$ and y-intercepts $\mu = 0$. From the data in the ?gure we deduce $(C = 2.7 \text{ tB} + 594.7 \text{ tB} = 33.40 \text{ tA} \text{ ? } 662.5$. These are used in obtaining the following results. (a) We ?nd $t \text{ ? } B \text{ ? } tB = 33.40 \text{ (t ? } A \text{ ? } tA) = 495 \text{ s}$ when $t \text{ ? } A \text{ ? } tA = 600 \text{ s}$. (b) We obtain $t \text{ ? } C \text{ ? } tC = 2.7 \text{ (t ? } B \text{ ? } tB) = 2.7$

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